

Induction of Heme Oxygenase Is a Rapid, Protective Response in Rhabdomyolysis in the Rat

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Abstract

Heme proteins such as myoglobin or hemoglobin, when released into the extracellular space, can instigate tissue toxicity. Myoglobin is directly implicated in the pathogenesis of renal failure in rhabdomyolysis. In the glycerol model of this syndrome, we demonstrate that the kidney responds to such inordinate amounts of heme proteins by inducing the heme-degradative enzyme, heme oxygenase, as well as increasing the synthesis of ferritin, the major cellular repository for iron. Prior recruitment of this response with a single preinfusion of hemoglobin prevents kidney failure and drastically reduces mortality (from 100% to 14%). Conversely, ablating this response with a competitive inhibitor of heme oxygenase exacerbates kidney dysfunction. We provide the first *in vivo* evidence that induction of heme oxygenase coupled to ferritin synthesis is a rapid, protective antioxidant response. Our findings suggest a therapeutic strategy for populations at a high risk for rhabdomyolysis. (*J. Clin. Invest.* 1992. 90:267–270.) Key words: kidney • crush syndrome • oxidants • hemoglobin • ferritin

Introduction

Organ damage may arise when heme proteins, such as myoglobin in muscle or hemoglobin in erythrocytes, escape from the intracellular space (1–7). Such heme-mediated organ injury underlies the syndrome of rhabdomyolysis, a disease instigated by muscle necrosis and dominated by kidney failure (1–7), and one that can be modeled in rats by the intramuscular injection of hypertonic glycerol (4, 5).

The exposure of the kidney in this syndrome to copious amounts of heme proteins led us to question whether heme oxygenase, the rate limiting enzyme in the degradation of heme (8), is induced in the kidney. In health this enzyme is mainly

expressed in organs involved in the clearance of effete erythrocytes (8); little enzyme activity is displayed by the intact kidney. This enzyme opens the heme ring, generates biliverdin, which is converted to bilirubin. Iron is released when the heme ring is opened, while carbon monoxide is liberated (8).

Interest in heme oxygenase has been recently stimulated by the speculation that heme oxygenase may be induced in tissues in response to oxidative stress (9, 10, 11). This thesis, so far untested *in vivo*, argues that the induction of heme oxygenase enables the removal of heme, a lipid-soluble, transmissible form of iron, as well as the generation of bilirubin, a metabolite with antioxidant properties (12). We reasoned that if induction of heme oxygenase represents an adaptive, protective response, then rhabdomyolysis assuredly provides a diseased state wherein the validity of this hypothesis can be tested. Moreover, we were stimulated by the observation that the intact kidney adapts to systemically administered hemoglobin by inducing heme oxygenase activity (13). We demonstrate that following rhabdomyolysis, rapid induction of heme oxygenase occurs in the kidney, and that the evocation of this response by hemoglobin before myonecrosis leads to a striking reduction in mortality. Our studies reveal the coupling of ferritin synthesis to enhanced heme oxygenase activity *in vivo*, a linkage thus far ignored *in vivo*, and one we suggest facilitates the safe sequestration of iron released as the heme ring is opened.

Methods

Glycerol model of rhabdomyolysis (glycerol acute renal failure [ARF]). This model was induced in male Sprague-Dawley rats (wt: 250–300 g) deprived of water, but allowed access to rat chow (Ralston Purina Co., St. Louis, MO) overnight. 15 h after the removal of water, rats were injected, under ether, with 50% glycerol in water, 10 ml/kg body wt, half the volume delivered into each anterior thigh muscle (4, 5).

Postischemic renal failure in the rat. This model was induced by right nephrectomy and left renal artery occlusion for 45 min under methohexital anesthesia (5 mg/100 g body wt given by the intraperitoneal route) (14).

RNA extraction and Northern analysis. RNA extraction, Northern analysis, and hybridization against cDNA probes were performed as detailed by Rosenberg et al. (15). The cDNAs to human heme oxygenase were obtained from Dr. Tyrrell (11), and the cDNAs to rat catalase and glutathione peroxidase were from Dr. Furata (16) and Dr. Yoshimura (17), respectively.

Determination of enzyme activity. Heme oxygenase activity was determined 6 h after the intramuscular injection of glycerol (10 ml/kg body wt) by two methods. We determined the generation of bilirubin by the colorimetric method (13). Kidney microsomes were incubated with hemin (17 μ M), hepatic cytosol (3 mg/ml), glucose-6-phosphate (2 mM), glucose-6-phosphate dehydrogenase (3 U/ml), and NADPH (0.4 mM). Using gas chromatography, we also determined carbon monox-

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1. Abbreviations used in this paper: ARF, acute renal failure.

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ide generation by kidney slices incubated overnight (18). Glutathione peroxidase and catalase activities were measured by the methods of Burk and Lawrence (19) and Aebi (20), respectively.

Renal studies: effect of prior induction of heme oxygenase. Rat hemoglobin, at a dose of 30 mg/100 g body wt in normal saline (62.5 mg/ml), was injected by tail vein 5 h before the commencement of overnight dehydration (13). Control rats received normal saline. 20 h after hemoglobin, all rats received an intramuscular injection of 50% glycerol, 10 ml/kg body wt and were subjected to daily serum creatinine determinations by a tail vein blood sample. In additional identically treated rats, kidneys were harvested for histology (14).

Renal functional studies: effect of inhibition of heme oxygenase. We employed a competitive inhibitor, tin protoporphyrin (21). Rats were deprived of food and water overnight for 15 h and injected the next morning with tin protoporphyrin, 20 μ mol/kg body wt in 0.5 ml of normal saline, or vehicle, subcutaneously. 3 h after this injection, rats were anesthetized with ether and injected with 50% glycerol, 7.5 ml/kg. Identical doses of tin protoporphyrin or vehicle were given 8 h and 24 h after the first dose.

Determination of ferritin content and mRNA. Ferritin content in kidney cytosol was determined by an ELISA method. Kidney cytosol was prepared by the method of Diez et al. (22). The primary antibody consisted of anti-rat ferritin obtained from Dr. Hamish Munro, Massachusetts Institute of Technology, Cambridge, MA (23). The secondary antibody consisted of anti-rabbit IgG alkaline phosphatase conjugate. The sensitivity of the assay was 100 pg/ml. cDNA probes for H-ferritin and L-ferritin were obtained from Dr. Hamish Munro.

All data are expressed as means \pm SE. The unpaired Student's *t* test was used for statistical comparisons between groups.

Results

Heme oxygenase mRNA was induced in the kidney, appearing 3 h after glycerol injection, attaining a 65-fold increase at 6 h (Fig. 1), and subsiding to a ninefold increase at 24 h. Enzyme activity was also increased at 6 h (Fig. 2). Increasing doses of glycerol (2.5–12.5 ml/kg body wt) incurred similar mRNA levels at 6 h.

We stimulated heme oxygenase by hemoglobin before provoking myolysis (activity before muscle injury: 0.228 \pm 0.037

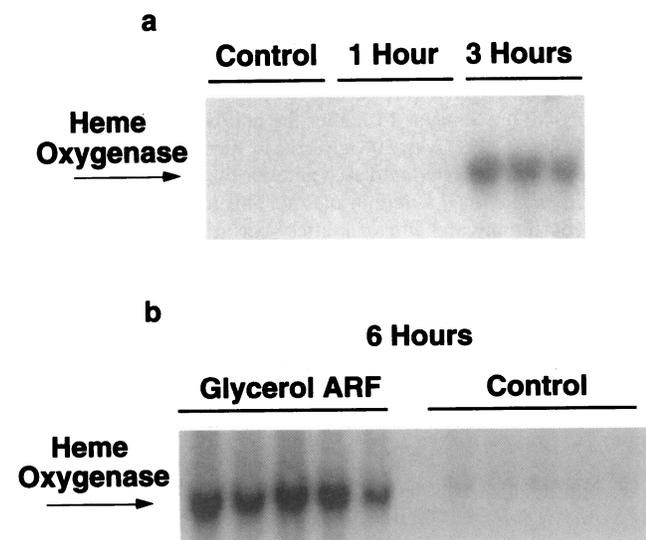


Figure 1. Heme oxygenase mRNA in rat kidney at 0, 1 and 3 h (a) and 6 h (b) after intramuscular injection of glycerol (10 ml/kg body wt). Each lane represents RNA from one kidney from an individual rat.

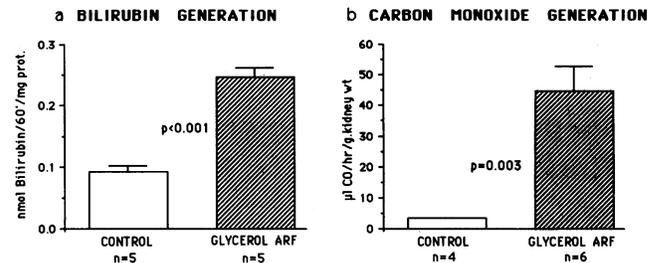


Figure 2. Heme oxygenase activity in rat kidney 6 h after the intramuscular injection of glycerol (10 ml/kg body wt) determined by bilirubin generation (a) and carbon monoxide generation (b).

vs. 0.085 \pm 0.003 nmol/mg protein per 60 min, $P < 0.05$). This exerted a striking protective effect (Fig. 3); rats pretreated with hemoglobin exhibited normal kidney function by the fifth day after myolysis, while all vehicle-treated rats died with kidney failure; mortality was reduced from 100% to 14% ($P = 0.002$ by Fischer's exact test). Hemoglobin pretreatment also reduced structural damage, as assessed by tubular necrosis and cast formation.

Porcine hemoglobin, similarly administered, also was protective (serum creatinine: 1.1 \pm 0.3 vs. 3.9 \pm 0.5 mg/dl, $P < 0.05$, and 0.9 \pm 0.3 vs. 6.6 \pm 0.2 mg/dl at 24 and 48 h after glycerol). Porcine hemoglobin, similarly administered 20 h previously, did not protect in the postischemic model (creatinine in treated and control rats: 2.9 \pm 0.5 vs. 2.2 \pm 0.5 mg/dl and 2.5 \pm 1.2 vs. 2.0 \pm 0.9 mg/dl, 24 and 48 h after reflow).

We validated the protective role of heme oxygenase by using a competitive inhibitor of heme oxygenase (Fig. 4). In this protocol we used a lower dose of glycerol (7.5 ml/kg) because of the mortality associated with a dose of 10 ml/kg. Tin protoporphyrin exacerbated kidney dysfunction in the glycerol model (Fig. 4). Tin protoporphyrin did not increase serum creatinine measured on two successive days in normal rats. Thus antagonizing the activity of heme oxygenase, induced in the kidney in this model, exacerbates kidney dysfunction.

Increased heme oxygenase activity in vitro engenders increased tissue content of ferritin (23, 24). We provide evidence of this same linkage in vivo. 6 h after myolysis, kidney content of ferritin was increased threefold (83.8 \pm 16.7 vs. 25.8 \pm 3.4, μ g/ml kidney cytosol, $P < 0.05$), at a time when ferritin mRNA was unaltered. However, 24 h after myolysis, mRNAs for both ferritin species are increased, sixfold and threefold for L-ferritin and H-ferritin, respectively (Fig. 5). That increased ferritin content may be protective in this model is suggested by studies in rats subjected to myolysis and treated with tin protoporphyrin: inhibition of heme oxygenase impairs kidney function and decreases ferritin content 24 h after myolysis (75.4 \pm 11.9 vs. 195.9 \pm 22.3, μ g/ml kidney cytosol, $P = 0.002$). Conversely, prior administration of hemoglobin is protective and increases kidney ferritin content measured just before glycerol (95.5 \pm 16.1 vs. 19.4 \pm 1.4, μ g/ml kidney cytosol, $P = 0.003$).

Other antioxidants, in contrast to heme oxygenase, were suppressed in this model. Activity was diminished for catalase (6 h, 0.20 \pm 0.02 vs. 0.27 \pm 0.01 and 24 h, 0.14 \pm 0.01 vs. 0.26 \pm 0.01, k/mg protein, $P < 0.01$) and for glutathione peroxidase (6 h, 67 \pm 3 vs. 121 \pm 8 and 24 h, 42 \pm 2 vs. 137 \pm 2 nmol/mg protein per min, $P < 0.001$). Such diminution in activity was accompanied by a decrease in mRNA (OD units) for catalase

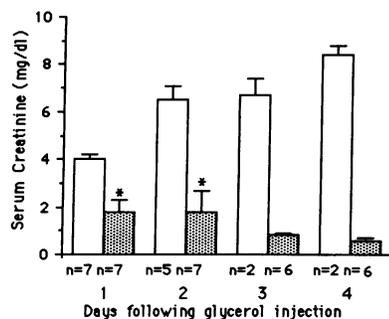


Figure 3. Functional significance of induction of heme oxygenase activity in glycerol-induced acute renal failure (10 ml/kg body wt): effect of prior induction of heme oxygenase. □, vehicle; ▨, rat hemoglobin; * $P < 0.005$.

(6 h, 495 ± 75 vs. 1389 ± 152 , $P < 0.001$, and 24 h, 275 ± 41 vs. 1179 ± 74 , $P < 0.01$), and a tendency for a decrement in glutathione peroxidase mRNA (6 h, 955 ± 77 vs. 1892 ± 420 , $P = 0.06$, and 24 h, 1615 ± 104 vs. 1950 ± 238 , $P = \text{NS}$).

Discussion

The induction of heme oxygenase that we observed following myolysis represents an adaptive rather than a maladaptive response in that inhibition of heme oxygenase worsens renal function, while, conversely, prior induction of heme oxygenase preserves renal function and reduces mortality.

The dominance of renal dysfunction in this syndrome (1, 2) led us to focus on the kidney. However, it is likely that other organs also display induction of heme oxygenase. Indeed the administration of hematin or hemoglobin induces hepatic heme oxygenase activity (25). Thus the reduction in mortality achieved with prior elicitation of this response reflects the accumulated salutary effects of induction in multiple organs. It must also be pointed out that induction of heme oxygenase in nonrenal organs such as the liver and spleen would facilitate the clearance of heme proteins and thus lessen the heme burden imposed on the kidney. Conversely, inhibition of heme oxygenase would impair extrarenal degradation, thereby augmenting the quantity of heme ultimately delivered to the kidney.

The renal toxicity of heme proteins is emphasized not only in studies of myolysis but also in investigations employing systemic infusions of heme proteins (1, 3, 4, 5, 6, 7). Such toxicity may arise from direct vasoconstriction (1) and from the propensity of heme proteins to precipitate in urinary casts as ferric heme, especially in the reduced pH range that occurs with renal acidification (6, 7). Heme proteins also induce injury through iron-catalyzed oxidative stress (4, 5). In this regard, the finding

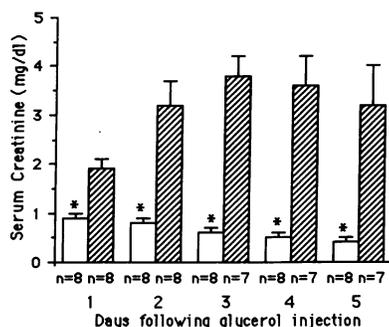


Figure 4. Functional significance of induction of heme oxygenase activity in glycerol-induced acute renal failure (7.5 ml/kg body wt): effect of inhibition of heme oxygenase. □, vehicle; ▨, tin protoporphyrin; * $P < 0.005$.

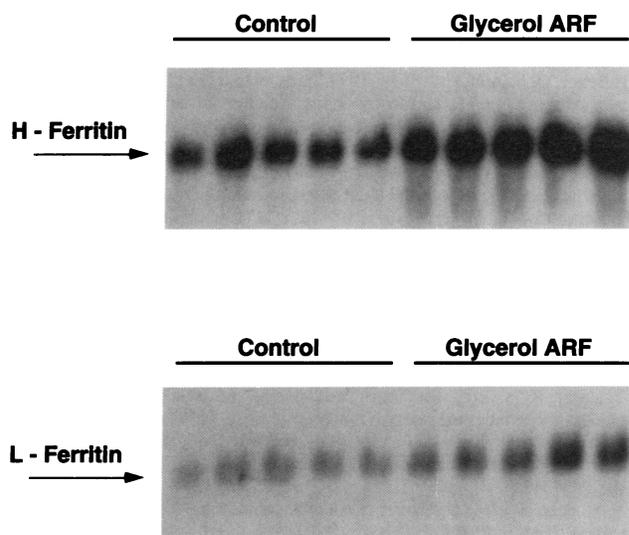


Figure 5. mRNA for L-ferritin and H-ferritin 24 h after the injection of intramuscular glycerol (10 ml/kg body wt).

that loading endothelial cells with hemin renders these cells exquisitely sensitive to small, otherwise innocuous doses of hydrogen peroxide, is perhaps relevant to the kidney (3).

Comparatively large amounts of hydrogen peroxide are generated during normal renal oxidative metabolism (26, 27); such generation of hydrogen peroxide may render the kidney particularly sensitive to heme toxicity. Minimizing the amount of heme to which the kidney is exposed, by prior induction of heme oxygenase, may thus lessen injury.

Induction of heme oxygenase was accompanied by decreased catalase and glutathione peroxidase activities. Such impairment may account for the hitherto unexplained findings of Guidet and Shah that the kidney generates increased amounts of hydrogen peroxide in this model (28).

Our findings are also germane to the phenomenon of resistance to acute kidney failure (29). Kidney dysfunction following myolysis is reduced by prior treatment with endotoxin or mercuric chloride; endotoxin and mercuric salts are inducers of heme oxygenase (8, 9). Our findings also suggest a therapeutic strategy in patients at an inordinate risk for the crush syndrome. Finally, our data may modulate the perception of the toxicity of heme: while toxic in large amounts, heme, in small doses, is strikingly protective *in vivo* against organ damage and mortality inflicted by overwhelming doses of heme itself.

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